# Carbon Tetrafluoride $\mathrm{CF}_{4}$ <br> Giga Class Grade: 99.999\% 

## Description

Carbon Tetrafluoride is a source of fluorine or carbon fluoride free radicals used in a variety of wafer etch processes. Carbon Tetrafluoride is used with oxygen to etch polysilicon, silicon dioxide, and silicon nitride. Carbon Tetrafluoride is relatively inert under normal conditions and is an asphyxiant. Under RF plasma conditions, the fluorine free radicals are typically in the form of $\mathrm{CF}_{3}$ or $\mathrm{CF}_{2}$. A higher purity Carbon Tetrafluoride results in superior control of the process, which results in better dimensional and profile control. Other halocarbons, as well as the presence of air or oxygen, are detrimental to the control of the anisotropic etch.

| Physical Constants |  |
| ---: | :--- |
| Chemical formula | $\mathrm{CF}_{4}$ |
| Molecular weight | 88.005 |
| Specific volume @ $+70^{\circ} \mathrm{F}\left(+21.1^{\circ} \mathrm{C}\right)$ | $4.396 \mathrm{ft}^{3} / \mathrm{lb}, 0.274 \mathrm{~m}^{3} / \mathrm{kg}$ |
| Critical pressure | $542.3 \mathrm{psia}, 37.39 \mathrm{bar}$ |
| Critical temperature | $-50.17^{\circ} \mathrm{F},-45.65^{\circ} \mathrm{C}$ |
| Specific gravity @ $70^{\circ} \mathrm{F}, 1 \mathrm{~atm}($ Air $=1)$ | 3.04 |


| Maximum Impurities |  |
| :---: | :---: |
| COMPONENT | CONCENTRATION (ppmv) |
| Carbon Dioxide | $<1.0$ |
| Carbon Monoxide | $<1.0$ |
| Moisture | $<1.0$ |
| Nitrogen | $<10.0$ |
| Oxygen | $<5.0$ |
| Other Halocarbons | $<1.0$ |
| Sulphur Hexafluoride | $<1.0$ |
| THC (as Methane) | $<1.0$ |


| Cylinder Specifications* |  | Contents | Pressure |  |  |
| :---: | :--- | :---: | :---: | :---: | :---: |
| Cylinder | Valve Outlet Options |  | Pounds | PSIG | BAR |
| 2 | CGA 580 | DISS 716 | 70 | 2000 | 139 |

*Additional cylinder sizes and/or valve outlets are available.

|  | www.spectragases.com |  | EL004, Aug 2004 |
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